

## Bioprospecting Unripe Pericarp of *Annona Reticulata* for the Presence of Bioactive Compounds

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### Abstract

The present work was carried out with the objective of isolating and identifying biologically significant chemical compounds from the ethanol extract of unripe pericarp of *Annona reticulata* through bioassay-guided fractionation. Ethanol extract of unripe pericarp was prepared and fractionation was done using silica gel chromatography. Thin-layer chromatography profiling was done and a single compound was obtained. The compound was subjected to various spectroscopic techniques for the elucidation of its structure. The result of spectral analysis of the compound was compared with the reported data which led to propose the structure of bioactive polyphenolic component as ellagic acid. Ellagic acid is reported to have antiproliferative and antioxidant properties. It is also marketed as dietary supplement against cancer, heart disease, and other medical problems.

**Keywords:** Bioprospecting, ellagic acid, ethanolic extract, spectroscopic techniques

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### INTRODUCTION

Medicinal plants and botanicals have always remained an important resource for the discovery of new drugs. Concept to product journey of such drugs is very challenging and involves various issues starting from bioprospecting to quality, safety, validation, and clinical studies. Bioprospecting has provided important leads to facilitate drug discovery from natural resources [1]. In an effort to expand the spectrum of biologically significant chemical compounds from natural resources, *Annona reticulata* belonging to Annonaceae was selected. *Annona reticulata* has been traditionally used in Indian folk medicines as vermifuge, anti-inflammatory agent, and in the treatment of diarrhea and dysentery. Various scientific studies reported its astringent, antianxiety, antistress, and spasmolytic activities [2].

The present work was carried out to isolate and identify biologically significant chemical compounds from the ethanolic extracts of unripe pericarp of *Annona reticulata* through bioassay-guided fractionation.

### MATERIALS AND METHODS

#### Preparation of Ethanol Extract of Unripe Pericarp

Mature unripe fruits of *Annona reticulata* used for the study were collected from Velimalai Hills, Kanyakumari district, Tamil Nadu, during the month of June to September. The pericarp of the fruits were separated from the seeds, cut into small pieces, shade dried, and powdered. The powder was soaked in ethanol for 48 hours and filtered [3].

#### Fractionation and Thin-Layer Chromatography

Fractionation was done using silica gel column chromatography [4]. Fractions were then collected, concentrated, and those having similar  $R_f$  value on the thin-layer chromatography (TLC) plates were pooled, numbered, and TLC profiling was done to get a single compound.

#### Spectroscopic Studies of Active Fraction

The isolated compound was subjected to various spectroscopic techniques for elucidation of the structure; namely, UV-visible spectral analysis, Infrared spectral analysis, <sup>1</sup>H

and  $^{13}\text{C}$ -Nuclear Magnetic Resonance (NMR) spectrum analysis and electrospray ionization mass spectrometry (ESI-MS) analysis.

## RESULTS AND DISCUSSION

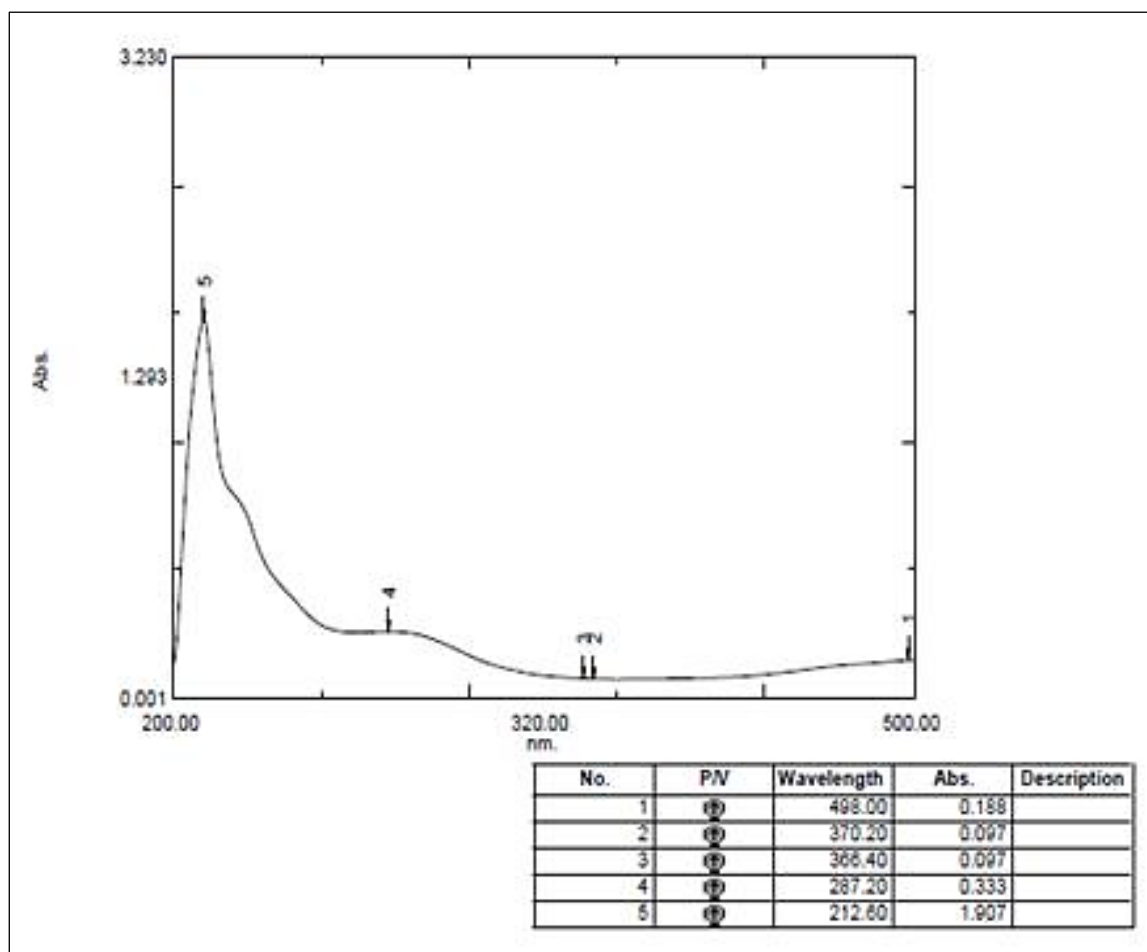
TLC of a single compound isolated from ethanol extract of unripe pericarp of *Annona reticulata* under UV light was conducted and a single spot was obtained with the fraction, the  $R_f$  value of which was determined as 0.64. The isolated compound was subjected to various spectroscopic techniques for the elucidation of the structure.

### *Spectral Analysis of Active Fraction of Ethanol Extract*

Figures 1 to 5 presents the results of spectral analysis carried out on the active fraction of ethanol extract. The UV spectrum of the active fraction showed absorption maxima at 253 nm and 366 nm. The infrared spectrum of the compound exhibited broad band in the range, 2924-3556  $\text{cm}^{-1}$  which is attributed to the -OH

stretching, while the band observed at 1697 $\text{cm}^{-1}$  corresponds to C=O stretching. The bands observed in the range, 1614-1510 $\text{cm}^{-1}$  are due to aromatic ring vibrations (C=C), while the ones at 1195  $\text{cm}^{-1}$  and 1055  $\text{cm}^{-1}$  are due to ester linkages (C-O). The band at 757  $\text{cm}^{-1}$  is assigned to aromatic C-H bending vibration. The ESI-MS spectrum showed a single strong peak at ( $m/z$ ) 301[M+H] $^+$  g/mol corresponding to ellagic acid.

$^1\text{H}$  NMR (DMSO, 500 MHz):  $^1\text{H}$  NMR spectrum of the active fraction showed a peak at 10.5 and 10.7 ppm which corresponds to -OH proton (4H, s). The peak at 7.46 ppm corresponds to aromatic C-H protons (2H). The peaks at 3.4 and 2.5 ppm correspond to a solvent proton, respectively.  $^{13}\text{C}$  NMR (DMSO, 126 MHz) spectroscopy showed  $\delta$  value at 159.14 (C-7, 7'), 148.11 (C-4, 4'), 139.55 (C-3, 3'), 136.39 (C-2, 2'), 112.31(C-1, 1'), 110.27 (C-5, 5') and 107.69 (C-6, 6'). The peaks from 38 to 40 ppm correspond to solvent peaks.



**Fig. 1:** UV Spectrum.

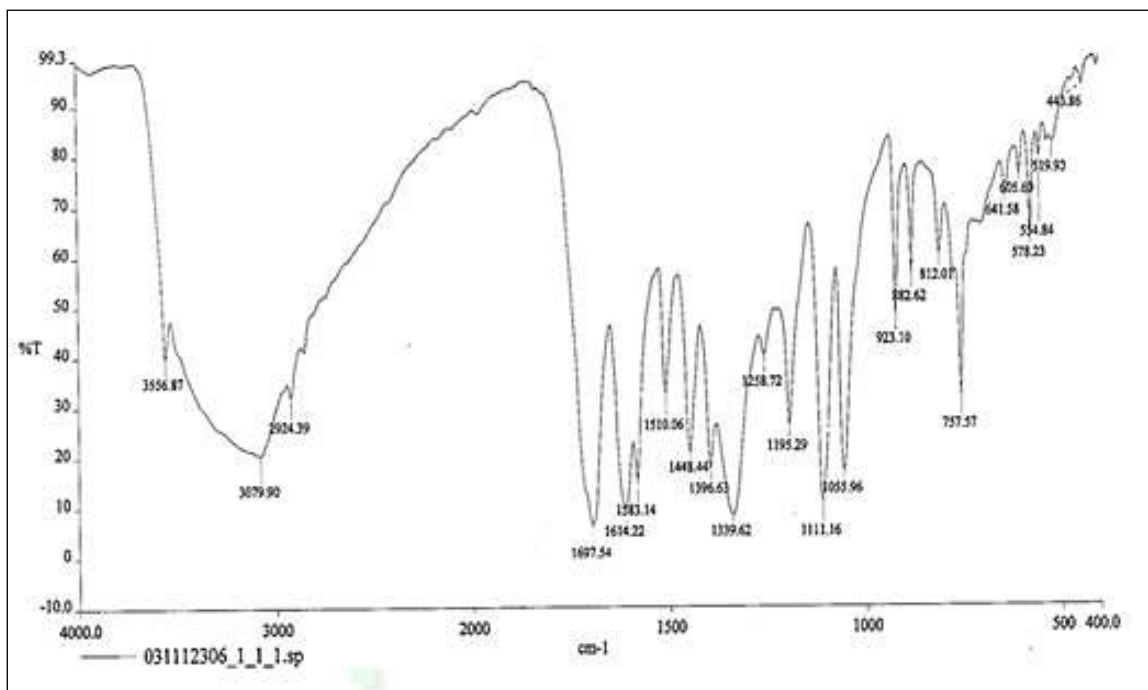


Fig. 2: Infrared Spectrum.

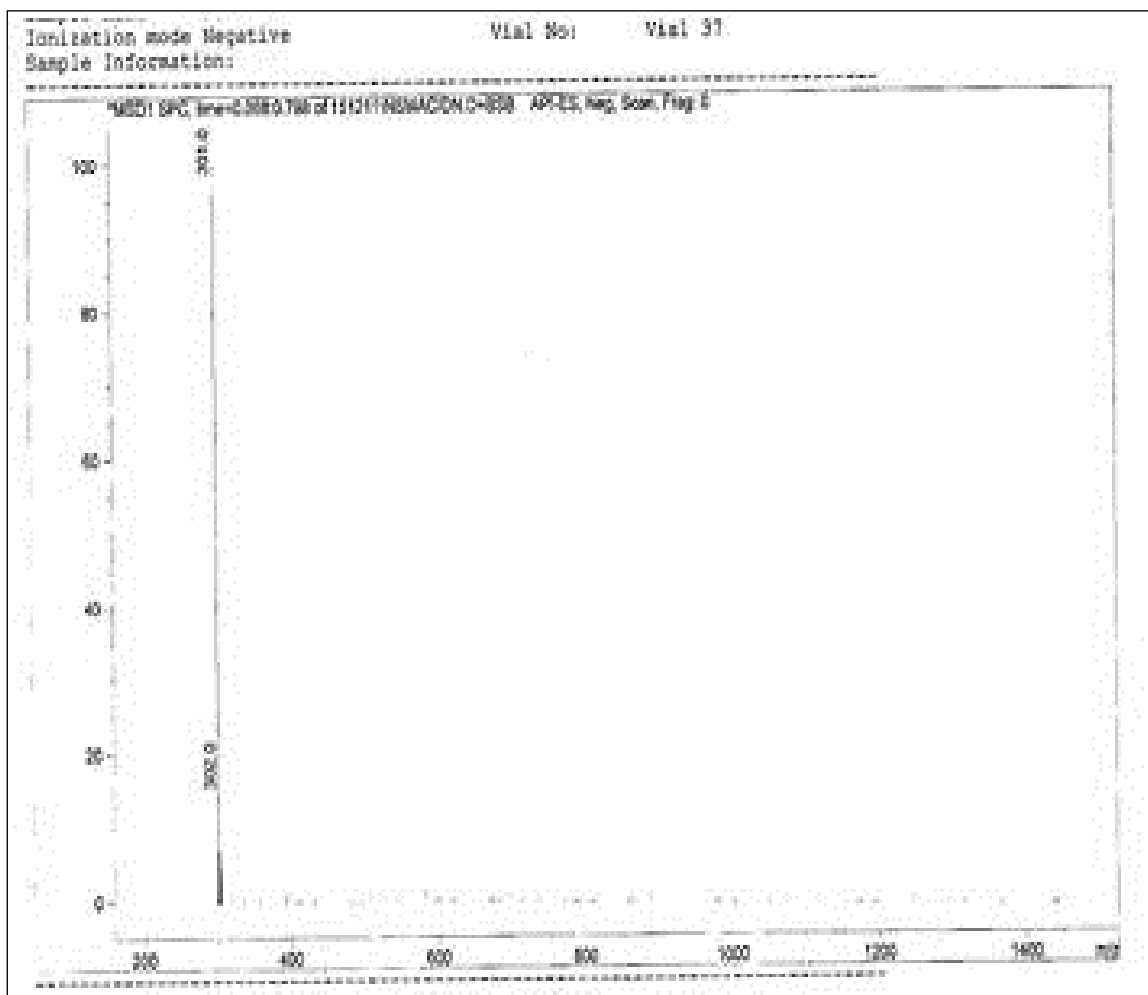
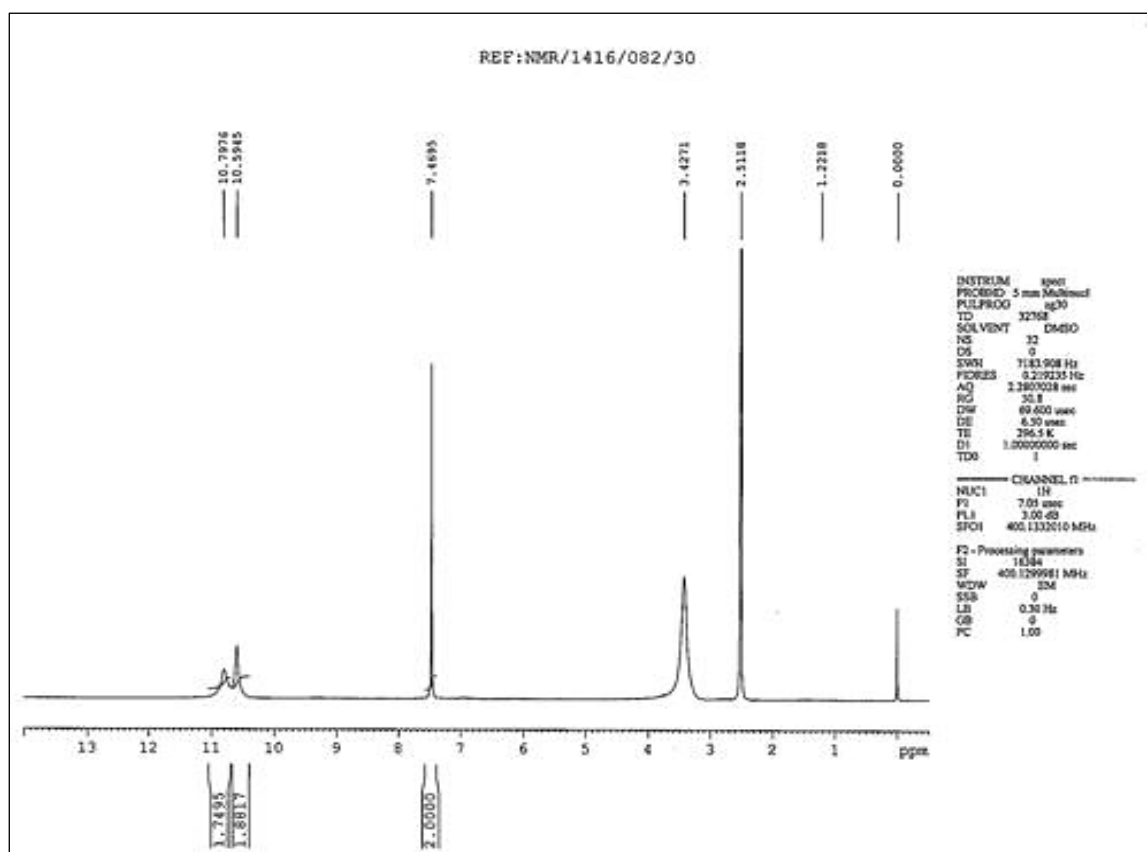
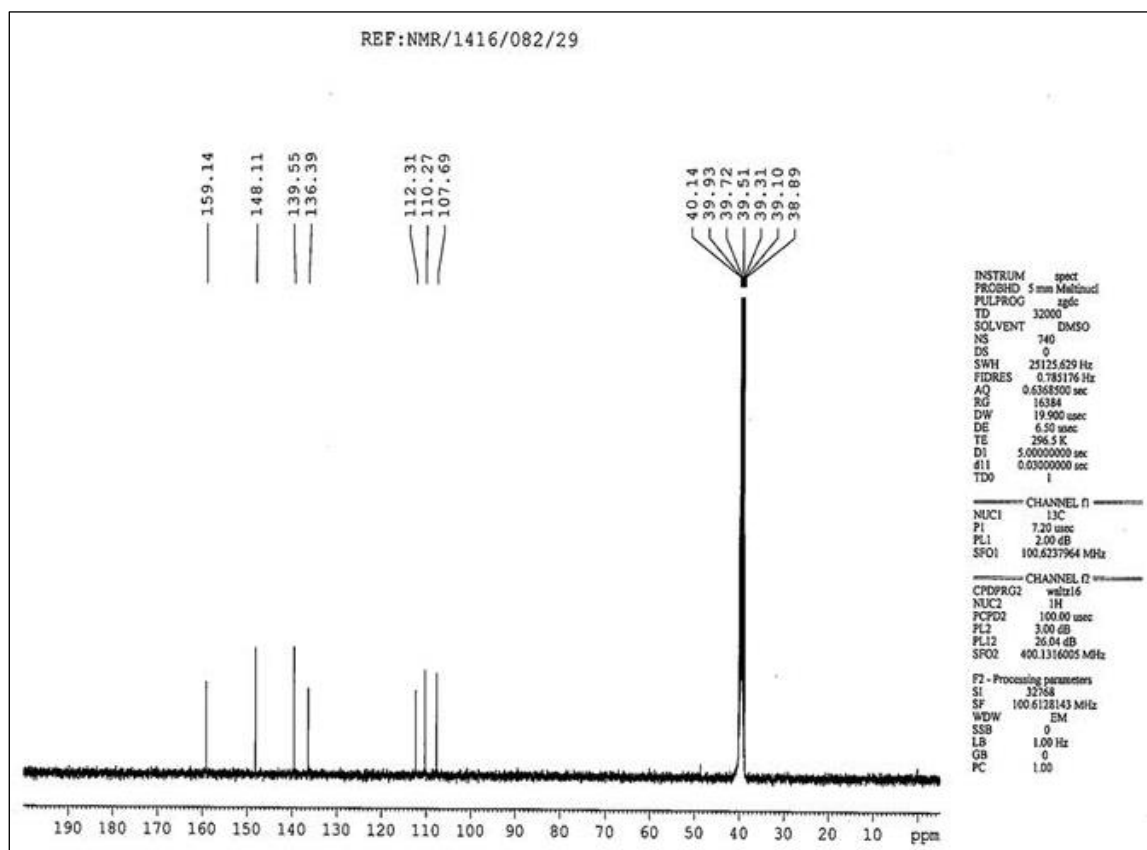


Fig. 3: ESI-MS Spectrum.

Fig. 4:  $^1\text{H}$  NMR Spectrum.Fig. 5:  $^{13}\text{C}$  NMR Spectrum.

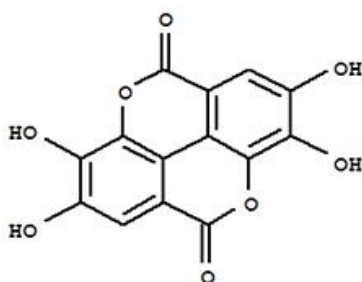


Fig. 6: Ellagic Acid.

The spectral analysis and the further comparison with the reported data led us to propose the structure of bioactive compound as ellagic acid (Molecular formula:  $C_{14}H_6O_8$ ), which is a polyphenolic compound (Figure 6).

Bioassay-guided fractionation has played an important role in the isolation of active compounds/fractions from natural products [5,6]. In the present study, bioassay-guided fractionation was employed to isolate the bioactive fractions/compounds from the unripe pericarp. The isolated fraction was subjected to ultraviolet, infrared,  $^1H$  NMR,  $^{13}C$  NMR and mass spectral analysis. Two aromatic ring proton(s) singlets d 7.45 ppm (2H, s, 4, 9- H) and four broad hydroxyl signals d 10.6 (2H, br) and 4.10 ppm (2H, br), which disappeared on deuterium exchange, were observed in its  $^1H$  NMR spectra. In its  $^{13}C$  NMR spectra, a set of seven carbon signals were observed. Two ppm indicated a carbonyl group, while the 148.1 to 107.7 ppm suggested the presence of a benzene ring including those for the 5-carbon units of the gallic acid skeleton. Thus, a symmetrical molecule was identified on the basis of its mass spectrum and NMR data. It was further confirmed as ellagic acid by comparison with the literature [7]. Thus, the results of spectral data suggest that the bioactive lead obtained was ellagic acid. The presence of ellagic acid has been reported in several medicinal plants [8,9,10,11]. Its presence in human dietary sources, especially fruits and nuts also has been reported [12,13,14].

Ellagic acid has been reported to have antiproliferative and antioxidant properties in a number of *in vitro* and small animal models and also prevents the destruction of p53 gene by cancer cells [15,16]. There are also reports on its chemoprotective effect in cellular models by reducing oxidative stress [17].

## CONCLUSIONS

The polyphenolic compound isolated from unripe pericarp of *Annona reticulata* identified as ellagic acid has been reported to be having anticancerous and antioxidant properties and hence the unripe pericarp can be used as a biological resource for producing drugs. The anticancerous activity and cytotoxicity of the ethanolic unripe pericarp extracts of *Annona reticulata* has to be screened using cancer cell lines. This might contribute to the development of valuable anticancer drug. Furthermore, the fractionated compound flashes more light into the pharmacological efficiency of the plant studied.

## CONFLICT OF INTEREST

Nil

## AUTHORS'S CONTRIBUTIONS

Mrs. V. S. Sangeetha carried out the study design and bioassay-guided fractionation of the extract and drafted the manuscript. Ms. Adheena Elza Johns helped in analyzing the spectral characteristics of the isolated compound. Dr. Beena Lawrence, the research guide participated in coordinating the drafted manuscript and helped to draft the final manuscript. All authors read and approved the final manuscript.

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